# CLASS PRACTICE DENSITY NOTES <br> **Use THIS to help you with your homework page. 

$$
\begin{array}{rr}
\text { volume }=\text { length } \times \text { width } \times \text { height } & \text { Label } \mathrm{cm}^{3} \\
\text { density }=\frac{m}{v} & \text { Label } \mathrm{g} / \mathrm{cm}^{3}
\end{array}
$$

1. An iron cube measures $5 \mathrm{~cm} \times 5 \mathrm{~cm} \times 5 \mathrm{~cm}$. What is its volume?
2. If the same iron cube's mass is 3.2 kg , what is its density in $\mathrm{g} / \mathrm{cm}^{3}$ ?
3. What is the density of a block of chocolate measuring $3 \mathrm{~cm} \times 4 \mathrm{~cm} \times 1 \mathrm{~cm}$, with a mass of $6 g$ ?
4. What is the density of a block of wood measuring $0.6 \mathrm{~cm} \times 3 \mathrm{~cm} \times 8 \mathrm{~cm}$ with a mass of $8.4 g$ ?
5. What has the greater density, a cube of water measuring $2 \mathrm{~cm} \times 2 \mathrm{~cm} \times 2 \mathrm{~cm}$ and having a mass of 8 g , or a block of plastic measuring $2 \mathrm{~cm} \times 3 \mathrm{~cm} \times 5 \mathrm{~cm}$ with a mass of 4 g ?
6. What is the density of a cube of paper measuring 7 cm on each side and its mass is 82 g ?

Water has a density of approximately $1 \mathrm{~g} / \mathrm{cm}^{3}$. In fact $1 \mathrm{~cm}^{3}$ of water used to be the standard for a gram. Objects will sink if their density is greater that water and will float if their density is less. For the following problems, decide if the block will sink or float.
7. A cube measuring 5 cm on each side and its mass is 25 g ; will it sink or float?
8. A block has a mass of 45 g and measures $4 \mathrm{~cm} \times 2 \mathrm{~cm} \times 4 \mathrm{~cm}$; will it sink or float?
9. A hollow iron cube has measures 4 cm on each side and has a mass of 30 g . Will the iron cube sink or float?
10. A cube made of very old hard wood, has a mass of 68 g and measures 9 cm a side, will it sink or float?

